



Information about the subject

Degree: Bachelor of Science Degree in Biotechnology

Faculty: Faculty of Veterinary Medicine and Experimental Sciences

Code: 1100209 **Name:** Thermodynamics and Kinetics

Credits: 6,00 **ECTS Year:** 2 **Semester:** 2

Module: Chemistry for Biomolecular Sciences

Subject Matter: Chemistry Physics **Type:** Compulsory

Department: -

Type of learning: Classroom-based learning

Languages in which it is taught: Spanish

Lecturer/-s:

1102

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Module organization

Chemistry for Biomolecular Sciences

Subject Matter	ECTS	Subject	ECTS	Year/semester
Chemistry	12,00	General Chemistry	6,00	1/1
		Organic Chemistry	6,00	1/2
Chemistry Physics	6,00	Thermodynamics and Kinetics	6,00	2/2

Learning outcomes

At the end of the course, the student must be able to prove that he/she has acquired the following learning outcomes:

- R1 The student has understood and assimilated the contents of the subject.
- R2 The student is able to solve problems or case studies related to the subject contents, by using different resources (bibliographic, IT, etc.).
- R3 The student is able to work in a laboratory, carrying out basic operations correctly and taking into account the corresponding safety standards. He/she understands the planning, development and purpose of the experience, and is able to contrast and validate the obtained results.
- R4 The student is able to write an intelligible and organized text on different aspects of the subject.
- R5 The student is able to present and defend his/her work adequately.
- R6 The student seeks bibliographic information from different sources and can analyze it with a critical and constructive spirit.
- R7 The student collaborates with the teacher and his/her peers throughout the learning process; he/she works in a team; treats everyone with respects, is proactive and fulfills the organization rules of the course.



Competencies

Depending on the learning outcomes, the competencies to which the subject contributes are (please score from 1 to 4, being 4 the highest score):

BASIC		Weighting			
		1	2	3	4
CB1	Students acquire and understand knowledge in their field of study based on general secondary education but usually reaching a level that, although supported on advanced text books, also includes aspects involving state-of-the-art knowledge specific to their area.		X		
CB2	Students are able to apply knowledge to their work in a professional way and have the competences enabling them to state and defend views and opinions as well as perform problem-solving tasks in their field of study.			X	
CB3	Students are able to collect and interpret relevant data (generally in their field of study) and give opinions that involve reflection on relevant social, scientific or ethical issues.			X	
CB4	Students can communicate information, ideas, problems and solutions to a specialized or non-specialized audience.		X		
CB5	Students develop the necessary learning skills to undertake further studies with a high level of autonomy.			X	

GENERAL		Weighting			
		1	2	3	4
CG01	Capacity to analyze and synthesize.				X

SPECIFIC		Weighting			
		1	2	3	4
CE22	Knowing and understanding contents, principles and theories related to biotechnology.				X



CE23	Knowing how to use laboratory equipment and to carry out basic operations for each discipline including: safety measures, handling, waste disposal and activity register.				X
CE24	Knowing basic and instrument laboratory techniques in the different areas of biotechnology.			X	
CE25	Knowing how to analyze and understand scientific data related to biotechnology.				X
CE29	Contrasting and checking results of biotechnological experimentation.			X	
CE30	Solving and analyzing problems posed by biotechnology.				X
CE31	Describing and calculating important variables of processes and experiments.				X

TRANSVERSAL		Weighting			
		1	2	3	4
CT02	Capacity to organize and plan.			X	
CT03	Mastering Spanish oral and written communication.			X	
CT05	Knowing and applying Basic ITC skills related to Biotechnology.			X	
CT06	Capacity to manage information (capacity to look for and analyze information coming from different types of sources).			X	
CT07	Problem solving.			X	
CT08	Decision making		X		
CT09	Capacity to work in interdisciplinary and multidisciplinary team.			X	
CT10	Interpersonal skills.		X		
CT11	Understanding multicultural and diverse environment	X			
CT12	Critical and self-critical capacity.				X
CT13	Ethics.		X		



CT14	Capacity to learn				X
CT15	Capacity to adapt to new situations			X	
CT16	Capacity to produce new ideas (creativity)				X
CT17	Leadership abilities	X			
CT18	Taking initiatives and enterprising spirit		X		
CT19	Capacity to apply theoretical knowledge			X	
CT20	Research skills				X
CT21	Sensitivity to environmental issues		X		

Assessment system for the acquisition of competencies and grading system

Assessed learning outcomes	Granted percentage	Assessment method
R1, R2, R4, R5	60,00%	Written test
R4, R5, R6, R7	20,00%	Submission of papers
R3, R5	20,00%	Laboratory test

Observations

The complete written test consists of a written test of theoretical content and a written test of practical content (problem solving). It is necessary to obtain a minimum of 4.5 out of 10 in each of them to average. You also need to score a minimum of 4.5 out of 10 on the Lab Test to average.



MENTION OF DISTINCTION:

According to Article 22 of the Regulations governing the Evaluation and Qualification of UCV Courses, the mention of "Distinction of Honor" may be awarded by the professor responsible for the course to students who have obtained, at least, the qualification of 9 over 10 ("Sobresaliente"). The number of "Distinction of Honor" mentions that may be awarded may not exceed five percent of the number of students included in the same official record, unless this number is lower than 20, in which case only one "Distinction of Honor" may be awarded.

Learning activities

The following methodologies will be used so that the students can achieve the learning outcomes of the subject:

- M1 Teacher presentation of contents, analysis of competences, explanation and in-class display of skills, abilities and knowledge.
- M2 Group work sessions supervised by the professor. Case studies, diagnostic tests, problems, field work, computer room, visits, data search, libraries, on-line, Internet, etc. Meaningful construction of knowledge through interaction and student activity.
- M3 Activities carried out in spaces with specialized equipment.
- M4 Supervised monographic sessions with shared participation..
- M5 Application of multidisciplinary knowledge.
- M6 Personalized and small group attention. Period of instruction and/or guidance carried out by a tutor to review and discuss materials and topics presented in classes, seminars, readings, papers, etc.
- M7 Set of oral and/or written tests used in initial, formative or additive assessment of the student
- M8 Group preparation of readings, essays, problem-solving, seminars, papers, reports, etc. to be presented or submitted in theoretical , practical and/or small-group tutoring sessions. Work done on the university e-learning.
- M9 Student's study: Individual preparation of readings, essays, problem-solving, seminars, papers, reports, etc. to be presented or submitted in theoretical, practical and/or small-group tutoring sessions. Work done on the university e-learning platform.



IN-CLASS LEARNING ACTIVITIES

	LEARNING OUTCOMES	HOURS	ECTS
ON-CAMPUS CLASS M1	R1, R2, R7	37,00	1,48
PRACTICAL CLASSES M2	R1, R4, R5, R6, R7	5,00	0,20
LABORATORY M3	R3	8,00	0,32
SEMINAR M4	R2, R7	3,00	0,12
GROUP PRESENTATION OF ASSIGNMENTS M5	R4, R5, R6, R7	3,00	0,12
TUTORIAL M6	R5, R7	2,00	0,08
ASSESSMENT M7	R1, R2, R3, R4, R5, R6, R7	2,00	0,08
TOTAL		60,00	2,40

LEARNING ACTIVITIES OF AUTONOMOUS WORK

	LEARNING OUTCOMES	HOURS	ECTS
AUTONOMOUS GROUP WORK M8	R3, R4, R5, R6, R7	18,00	0,72
AUTONOMOUS INDIVIDUAL WORK M9	R1, R2, R3, R4, R5, R6	72,00	2,88
TOTAL		90,00	3,60



Description of the contents

Description of the necessary contents to acquire the learning outcomes.

Theoretical contents:

Content block	Contents
Thermochemistry	Basic thermodynamic concepts and Zero Principle: Temperature First and Second Principles: Enthalpy, Entropy, and Free Energy Thermodynamics and Equilibrium: Laws of chemical combinations. Chemical Species: Stoichiometry and Chemical Equilibrium Phase transitions Colligative properties
Chemical kinetics	Kinetics of chemical reactions. Chemical catalysis. Introduction to biocatalysis. Transportation phenomena.
Laboratory	Determination of the differential heat of dissolution of a solute. Discoloration reaction kinetics in phenolphthalein in basic medium. Catalysis. Kinetic study of the hydrolysis of an ester. Cryoscopy. Determination of the molecular mass of a solute.



Organization of the practical activities:

	Content	Place	Hours
PR1.	Determination of the differential heat of dissolution of a solute.	Laboratory	2,00
PR2.	Discoloration reaction kinetics in phenolphthalein in basic medium.	Laboratory	2,00
PR3.	Catalysis.	Laboratory	2,00
PR4.	Kinetic study of the hydrolysis of an ester. Cryoscopy. Determination of the molecular mass of a solute.	Laboratory	2,00
PR5.	Review of calculations and procedures, collaborative work	Lecture room	5,00

Temporary organization of learning:

Block of content	Number of sessions	Hours
Thermochemistry	18,00	36,00
Chemical kinetics	8,00	16,00
Laboratory	4,00	8,00



References

Fisico-Química. Atkins P. W. Ed. Addison-Wesley Iberoamericana, 1986
Química Física. Díaz-Peña M., Roig Muntaner. Ed. Alambra, 1983
Manual de Termodinámica. Gandía V. Ed. Universitat de Valencia, 1995
Fisicoquímica para Farmacia y biología. Sanz Pedrero, P. Ed. Masson-Salvat, 1992
Problemas de Fisicoquímica. Levine I. N., Ed. McGraw-Hill, 2005
Introducción a la Fisicoquímica. J.L. Moreno Frigols, R.García Doménech, G.Antón
Fos.Universitat de València, 2011

Química general. R.H. Petrucci, W.S. Harwood y F. G. Herring. Ed. Prentice. Hall, 2003.
Química. R. Chang. Ed. Mc Graw-Hill, 1997.
Química principios y reacciones. W. I.Masterton y C.N. Hurley. Ed. Thomson. Paraninfo, 2003.
Fundamentos de fisicoquímica. S.H.Maron y C.F. Prutton. Ed. Limusa, 1998.
Química curso universitario. B.M. Mahan y R.J. Myers.Ed. Addison-Wesley Iberoamericana,1990.
Química general. J.B. Umland y J.M. Bellana, Ed.Thomson, 1999.



Addendum to the Course Guide of the Subject

Due to the exceptional situation caused by the health crisis of the COVID-19 and taking into account the security measures related to the development of the educational activity in the Higher Education Institution teaching area, the following changes have been made in the guide of the subject to ensure that Students achieve their learning outcomes of the Subject.

Situation 1: Teaching without limited capacity (when the number of enrolled students is lower than the allowed capacity in classroom, according to the security measures taken).

In this case, no changes are made in the guide of the subject.

Situation 2: Teaching with limited capacity (when the number of enrolled students is higher than the allowed capacity in classroom, according to the security measures taken).

In this case, the following changes are made:

1. Educational Activities of Onsite Work:

All the foreseen activities to be developed in the classroom as indicated in this field of the guide of the subject will be made through a simultaneous teaching method combining onsite teaching in the classroom and synchronous online teaching. Students will be able to attend classes onsite or to attend them online through the telematic tools provided by the university (videoconferences). In any case, students who attend classes onsite and who attend them by videoconference will rotate periodically.

In the particular case of this subject, these videoconferences will be made through:

☒ Microsoft Teams

☐ Kaltura



Situation 3: Confinement due to a new State of Alarm.

In this case, the following changes are made:

1. Educational Activities of Onsite Work:

All the foreseen activities to be developed in the classroom as indicated in this field of the guide of the subject, as well as the group and personalized tutoring, will be done with the telematic tools provided by the University, through:

☒ Microsoft Teams

☐ Kaltura

Explanation about the practical sessions:

The sessions of the laboratory practices will be carried out in a non-face-to-face way, using Teams and UCVNet tools, through seminars on the theoretical foundation and data processing.



2. System for Assessing the Acquisition of the competences and Assessment System

ONSITE WORK

Regarding the Assessment Tools:

☒ The Assessment Tools will not be modified. If onsite assessment is not possible, it will be done online through the UCVnet Campus.

☐ The following changes will be made to adapt the subject's assessment to the online teaching.

Course guide		Adaptation	
Assessment tool	Allocated percentage	Description of the suggested changes	Platform to be used

The other Assessment Tools will not be modified with regards to what is indicated in the Course Guide.

Comments to the Assessment System: